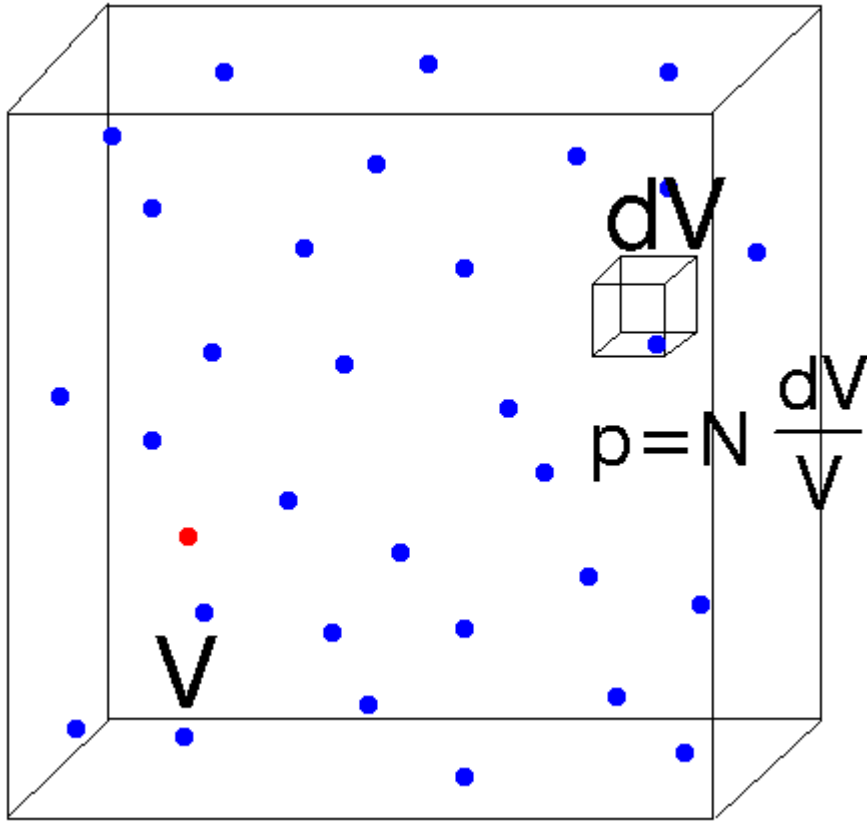
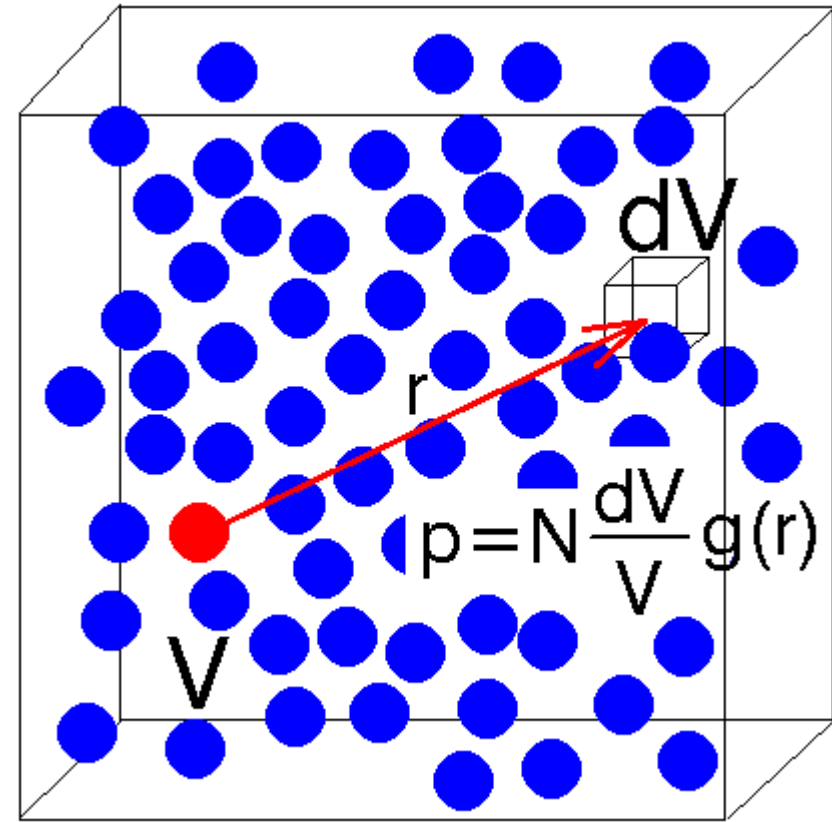


Correlation functions:

- radial distribution function (also pair correlation/distribution function)
= probability of finding a particle at distance r (from another particle), normalized to ideal gas
- structure factor (diffraction \rightarrow Fourier transform)
- angular correlation function – good for small nonspherical molecules
- time autocorrelation functions

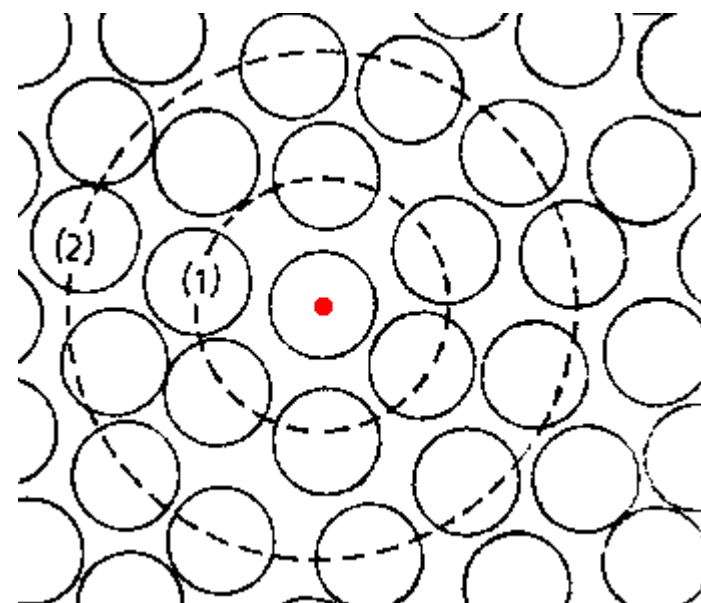
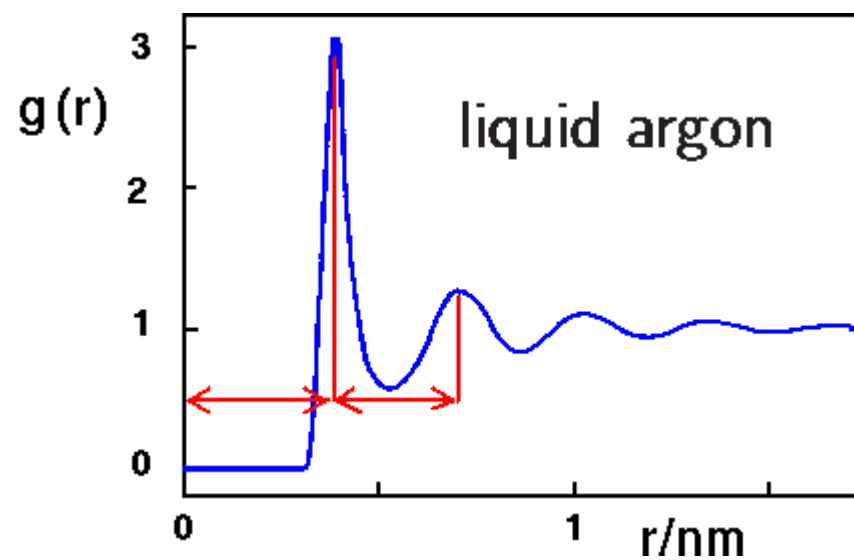
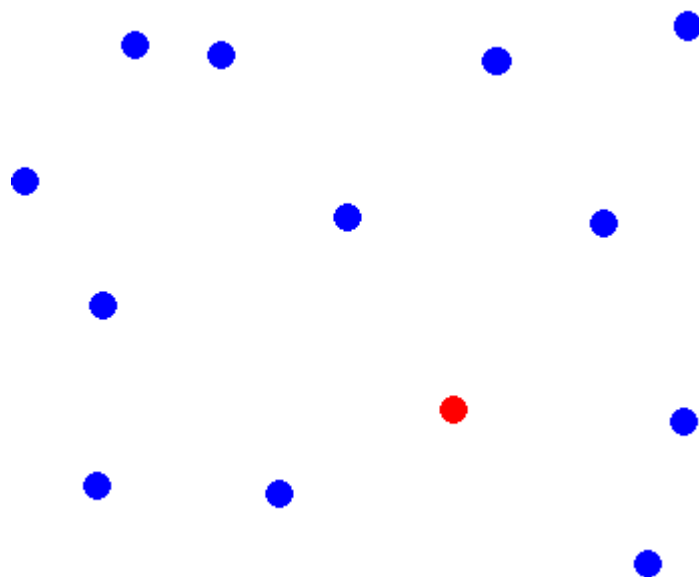
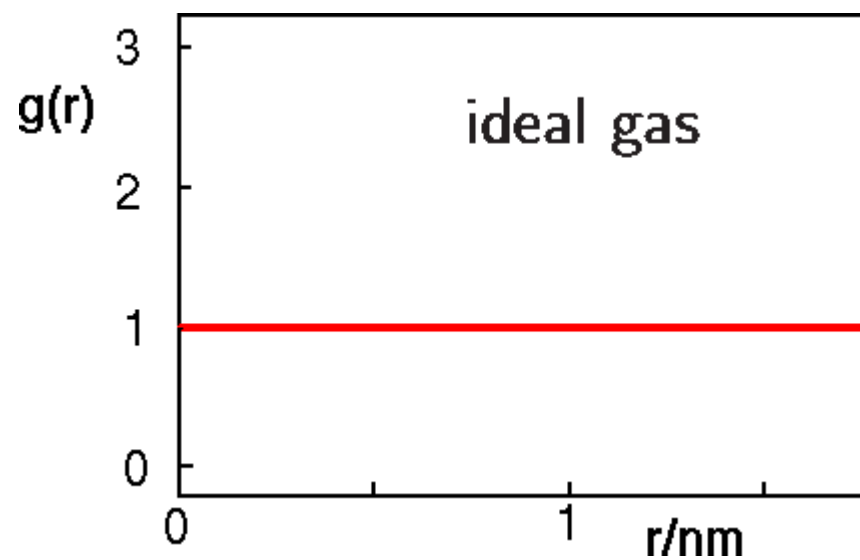


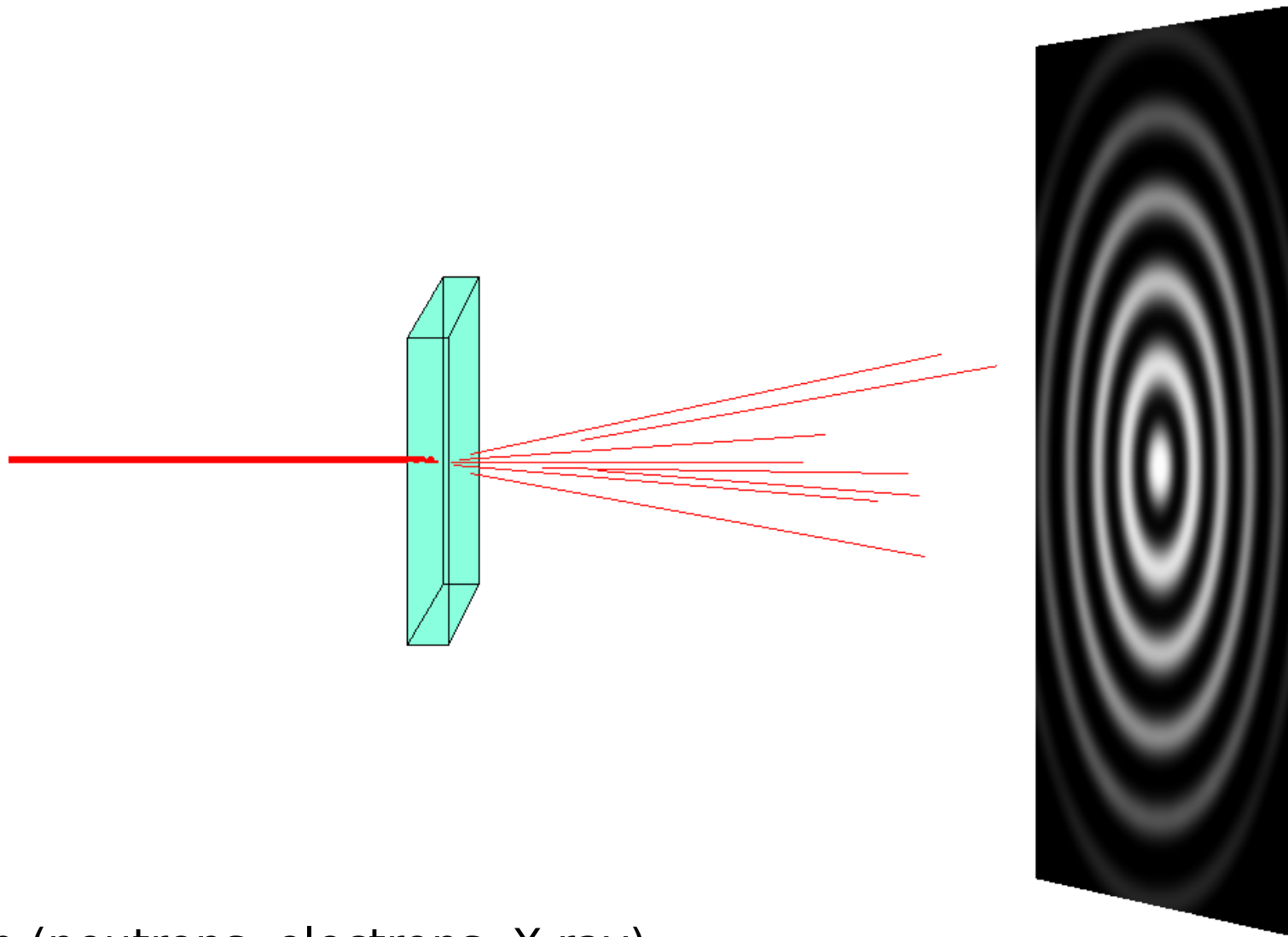
randomly distributed molecules
(ideal gas)



liquid

$g(r)$ = pair correlation function = radial distribution function = probability density of finding a particle r apart from another particle, normalized so that for randomly distributed particles (ideal gas) it is 1

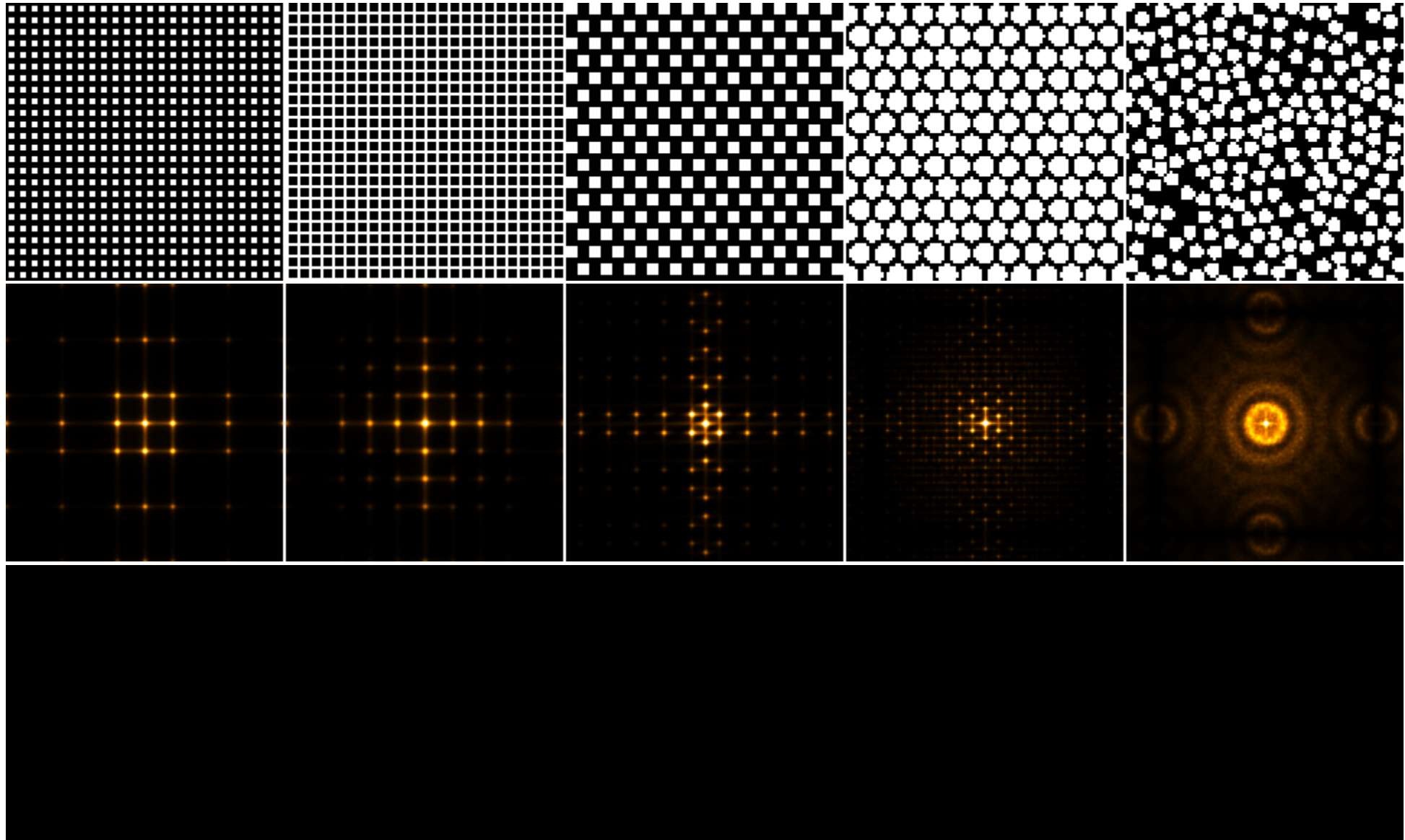




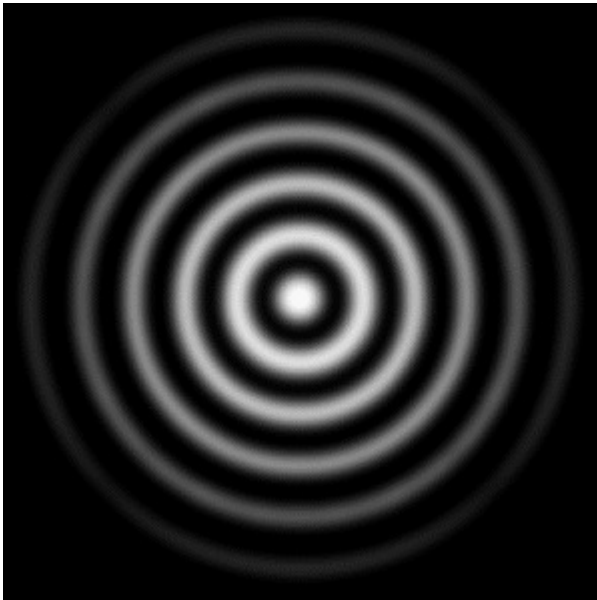
- Diffraction (neutrons, electrons, X-ray)
⇒ “structure factor”

How to obtain structure

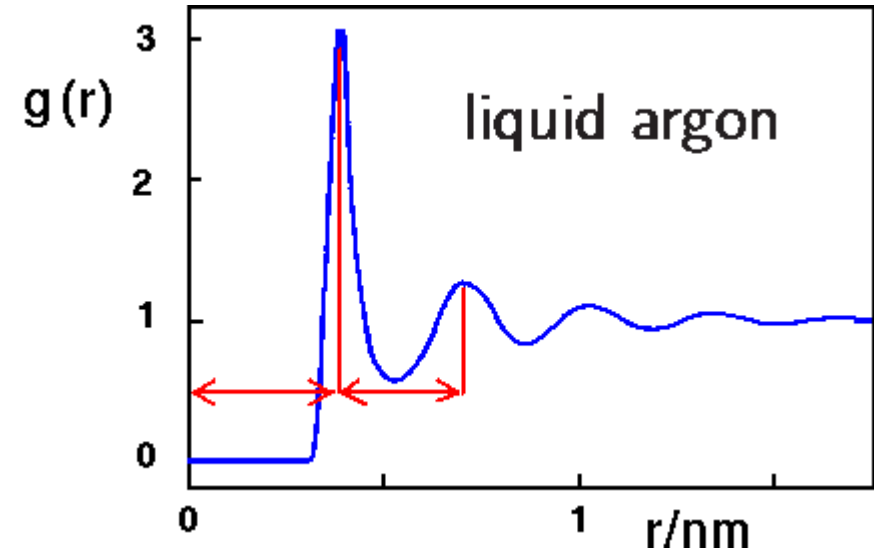
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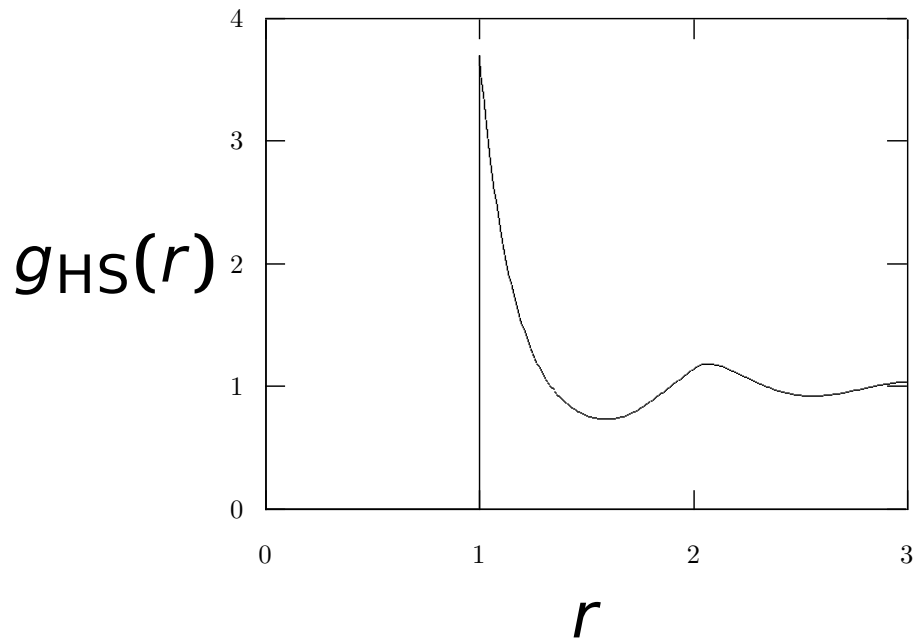
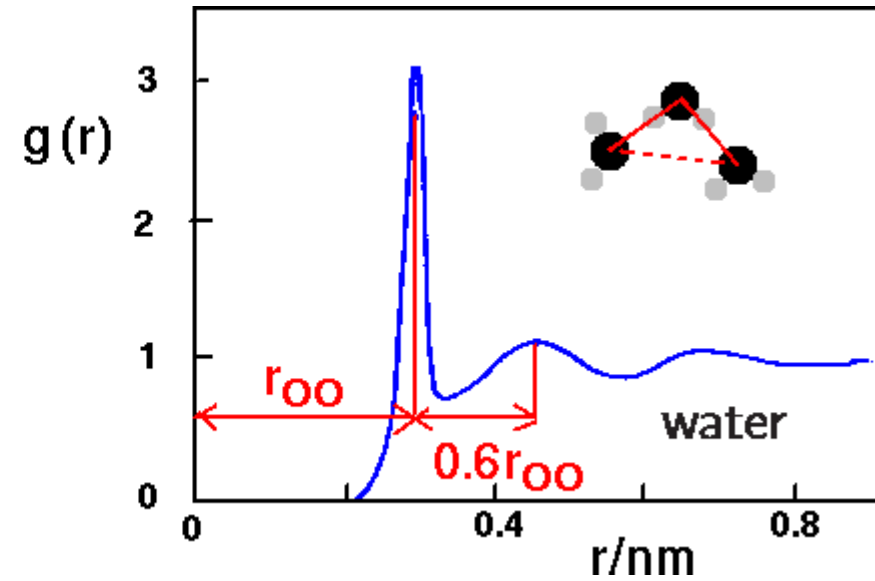
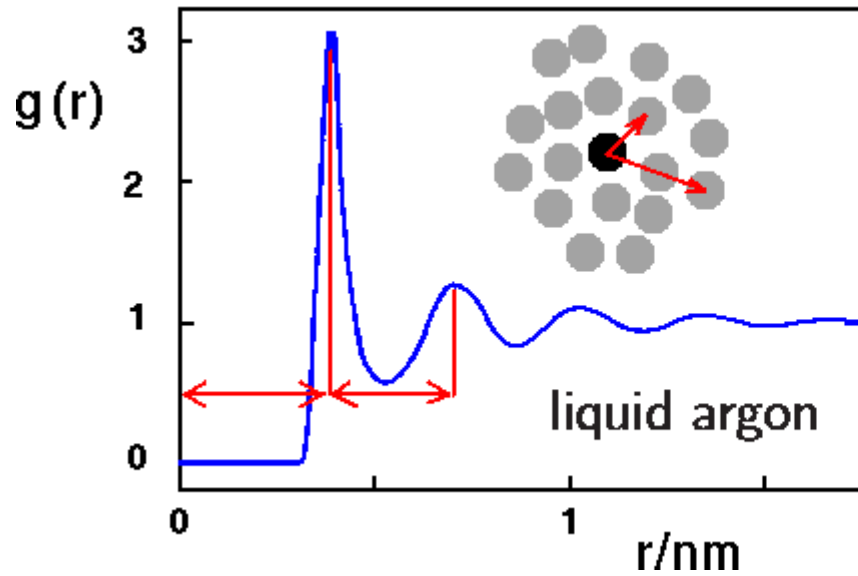


Correlation function from structure factor



inverse
Fourier
transform



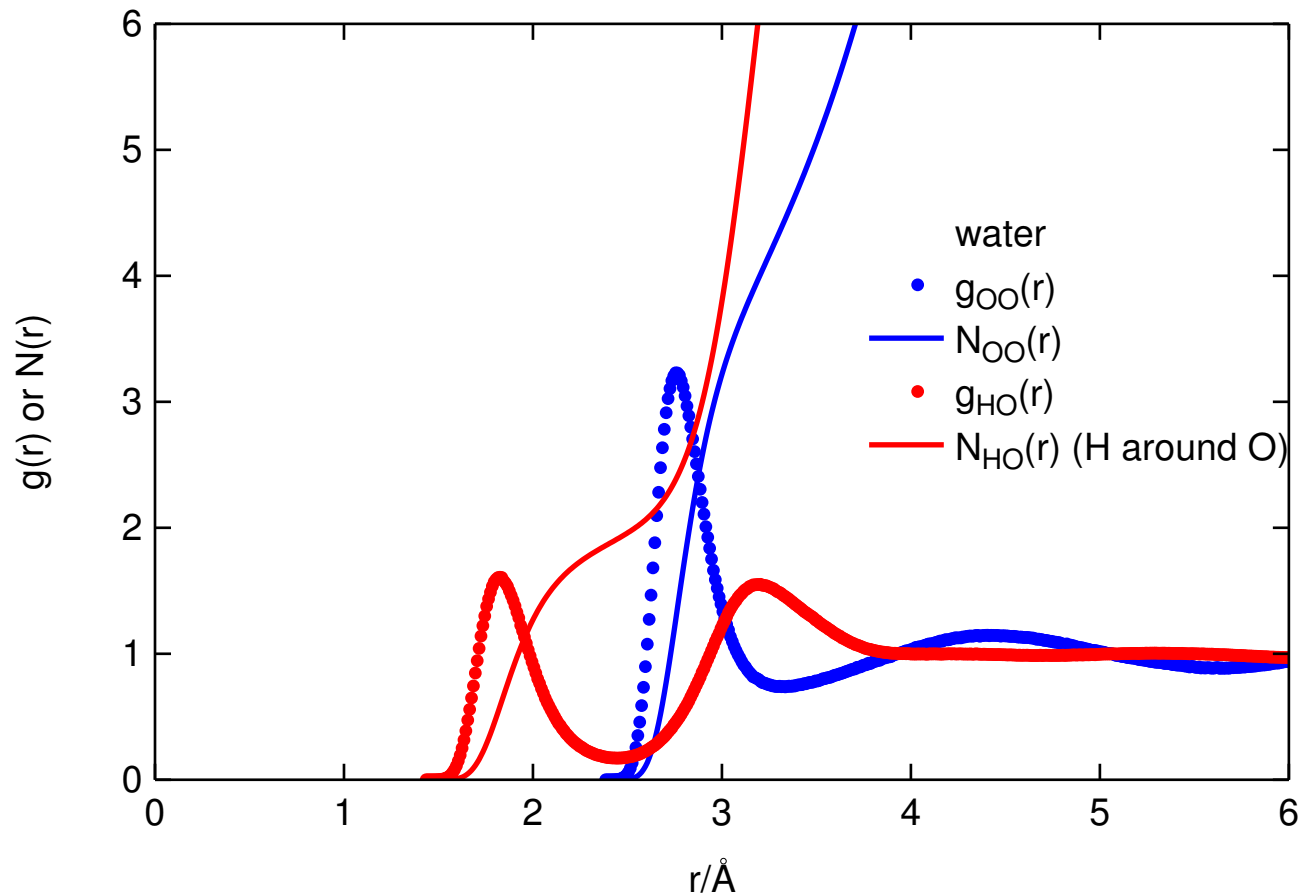


- The structure of simple fluid (argon, HS) is organized by shells.
- The structure of water is determined by the tetrahedral geometry of hydrogen bonds.
- After several molecular diameters, the correlations decay to zero.

Also “cumulative radial distribution function”

$$N(r) = 4\pi\rho \int_0^r g(r')r'^2 dr'$$

For r_{\min} = first minimum on the RDF curve, $N(r_{\min})$ = “coordination number” = averaged number of molecules in the first shell



Histogram of the number of particle pairs, \mathcal{N}_i , so that

$$r \in [r_i - \Delta r/2, r_i + \Delta r/2) \quad \text{alternatively} \quad \mathcal{I}_i = [r_i, r_i + \Delta r)$$

The volume of the shell

$$\Delta V_i = \frac{4\pi}{3} \left[\left(r_i + \frac{\Delta r}{2} \right)^3 - \left(r_i - \frac{\Delta r}{2} \right)^3 \right]$$

Mean number of molecules around a selected particle in case of uniformly distributed molecules (ideal gas, $\rho = N/V$):

$$\frac{1}{2} \rho \Delta V_i$$

Sum over all particles (1/2 to count each pair just once):

$$\mathcal{N}_i^{\text{id. gas}} = \frac{N}{2} \rho \Delta V_i$$

Radial distribution function:

$$g(r_i) = \frac{\langle \mathcal{N}_i \rangle}{\mathcal{N}_i^{\text{id. gas}}} = \frac{2 \langle \mathcal{N}_i \rangle}{N \rho \Delta V_i}$$

3D (e.g., in periodic b.c.):

$$g(r) \equiv g(r_{12}) = \frac{N(N-1)}{\rho^2 Q_{\text{NVT}}} \int \dots \int \exp[-\beta U(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)] d\vec{r}_3 \dots d\vec{r}_N$$

Equivalently

$$g(r) = \left(1 - \frac{1}{N}\right) V \langle \delta(\vec{r}_{12} - \vec{r}) \rangle$$

For a mixture:

$$g_{ij}(r) = V \langle \delta(\vec{r}_{12} - \vec{r}) \rangle$$

Normalization (fluid):

$$\lim_{N \rightarrow \infty, r \rightarrow \infty} g(r) = 1$$

NB: ideal gas at finite N : $g(r) = 1 - 1/N$ (e.g., in periodic b.c.)

Number of particles around one chosen particle (in NVT):

$$\int_V \rho g(\vec{r}) d\vec{r} = N - 1$$

Histogram of the count of pairs of particles, \mathcal{N}_i , so that $r \in \mathcal{I}_i$

$$\mathcal{I}_i = [r_i - \Delta r/2, r_i + \Delta r/2), \quad \text{optionally } \mathcal{I}_i = [r_i, r_i + \Delta r)$$

$$r_i = i\Delta r, \quad i = 1, \dots, i_{\max}$$

$$\begin{aligned} \langle \mathcal{N}_i \rangle &= \frac{1}{Q_{\text{NVT}}} \sum_{j < k} \int_{r_{jk} \in \mathcal{I}_i} \exp[-\beta U(\vec{r}^N)] d\vec{r}^N \\ &= \frac{1}{Q_{\text{NVT}}} \binom{N}{2} V \int_{r_{12} \in \mathcal{I}_i} \left\{ \int \exp[-\beta U(\vec{r}^N)] d\vec{r}_3 \dots d\vec{r}_N \right\} d\vec{r}_{12} \\ &= \frac{N}{2} \rho \int_{\mathcal{I}_i} g(r) d\vec{r} \\ &\approx \frac{N^2}{2V} g(r_i) \Delta V_i \end{aligned}$$

The formula again:

$$g(r_i) = \frac{2 \langle \mathcal{N}_i \rangle}{N \rho \Delta V_i}$$

For simple fluid with pair forces only:

Residual internal energy:

$$\langle U \rangle = \frac{1}{Q_{\text{NVT}}} \int \sum_{i < j} u_{ij}(r_{ij}) e^{-\beta U} d\vec{r}_1 \dots d\vec{r}_N$$

$$\langle U \rangle = \binom{N}{2} \frac{V}{Q_{\text{NVT}}} \int e^{-\beta U} 4\pi r_{12}^2 dr_{12} u(r_{12}) d\vec{r}_3 \dots d\vec{r}_N$$

$$\langle U \rangle = \frac{N}{2} \rho \int u(r) g(r) d\vec{r} = 2N\pi\rho \int u(r) g(r) r^2 dr$$

Pressure:

$$\frac{\beta P}{\rho} = 1 - \frac{2\pi}{3} \beta \rho \int g(r) u'(r) r^3 dr$$